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~ Elements within the same family of  
the periodic table have simi(ar  
properties because they have the same  
number of valence electrons. M  
Vocabulary Skills: Applying Definitions  
On the periodic table below, label each  
of the following: acndoid series alkaline  
earth ...

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10—2 Review and Reinforcement (continued) 9.  $0.0621 \text{ mol SS} = (Q \$ \text{gK2S}$  10.  $65.8\text{gMg} = \text{l. atoms Mg}$  Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in  $0.40 \text{ mol}$  of sulfur- 12. Calculate the number of atoms in  $2.30 \text{ mol}$  of silver. G.oux 13. You have a tank of NOZ gas.

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3—1 Review and Reinforcement Early Models of the Atom On the line at the left, write the letter of the chemist who proposed each of the ideas listed below. Each letter may be used once, more than once, or not at all. ... 17. 12 protons

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and 10 electrons 18. 8 protons and 10 electrons

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'16—1A Review and Reinforcement, pages 9-10 6. Answers Will Vary. When the rates Of forward and reverse reactions are equal, equilibrium is established. For the example of a liquid placed in. a container, equilibrium is established when the Equid vaporizes at

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the same rate that vapor condenses. 7, NO, in reversible reactions at equilibrium,

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11—1 Review and Reinforcement  
(continued) Solve each of the following problems as directed. Show all your work. 13. How many moles of magnesium are required to react with 2.0 mol of hydrochloric acid? The ... 17. Zinc reacts with  $H_2SO_4$  to yield  $ZnSO_4$  and hydrogen gas. How many moles of hydrogen will

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11—1 Review and Reinforcement  
Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase may be used once, more than once, or not at all reactants molar ratio quantitative actual particles. condensation of matter coefficients subscripts mass

## **10-2 Review and Reinforcements - Front Page**

17. The volume of gas particles



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themselves is assumed to be \_\_\_\_\_. 18. Gas molecules are said to be in \_\_\_\_\_. 19. The collisions between gas particles are \_\_\_\_\_. 20. The temperature of a gas is a measure of the average \_\_\_\_\_ of the gas ... 12-1 Review and Reinforcement

...

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1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed ... 17. distillation 18. A magnet could be used to separate them because iron is attracted to a magnet but

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