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but during a physical change, the composition of matter . 13. Complete the following table about chemical changes. Chemical Changes Type of Change Are New Substances Formed? Example Chemical Yes Physical Sugar dissolving in water Name \_\_\_\_ Class \_\_\_\_ Date \_\_\_\_ Chapter 2 Properties of Matter

### Chapter 2: Properties of Matter - Practice Test Questions ...

< Back Next > Preview Main Chapter 2 Section 1 What Is Matter? •Describe the two properties of all matter. •Identify the units used to measure volume and mass. •Compare mass and weight. •Explain the relationship between mass and inertia. Objectives

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Properties of Matter Chapter 2 WordWise. Solution. A mixture that results when substances dissolve to form a homogeneous mixture. Compound. A substance that can be broken down into two or more simpler substances. Physical Change. A change which the composition of matter stays the same.

**Chapter 2 Properties of Matter Section 2.3 Chemical Properties**

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Chapter 2 Properties of Matter Section 2.1 Classifying Matter (pages 38-44) This section explains how materials are classified as pure substances or mixtures. It discusses types of pure substances and mixtures. Reading Strategy (page 38) Summarizing As you read, complete the classification of matter in the diagram below.

**Unit 1 : Matter and Energy : Chapter 2. Properties of Matter**

Interactive Reader 29 Matter SECTION2 Properties of Matter Matter Name Class Date CHAPTER 2 READING TOOLBOX Underline As you read, use a red pen or colored pencil to underline physical properties of matter. Use a blue pen or colored pencil to underline chemical properties of matter. EHHDBG@ <EHL>K 1. Identify List three proper-ties of objects ...